Chapter 2

Elements and Compounds

LESSON PLAN

SPECIFIC OBJECTIVES

The students will learn about

- matter and its parts
- elements, their classification; atoms and symbol of an element
- compounds and their properties
- molecule; molecule of an element and compound; formula of a compound

Teaching Aids

Pictures/charts/models/animation on matter, its types; some examples of metals and nonmetals; structure of atom; symbols of elements (old, new as given in the chapter) compounds of some elements; electrolysis of water; chemical formulae of some compounds (as given in the chapter)

Teaching Strategy

- Teacher will define matter and its categories, i.e., pure substances like elements and compounds, and impure substances like mixtures.
- Teacher will also define atoms and molecules.
- Now, teacher will explain elements, their classifications, i.e., metals, nonmetals and metalloids.
- Teacher will discuss properties of metals, nonmetals and metalloids.
- Teacher will explain structure of atom of an elements.
- Now, teacher will ask students to solve 'Check Point 1'.
- Teacher will discuss the symbols of elements.
- Teacher will explain the concept of compounds, their properties and constituents of compounds and will demonstrate Activities 1 and 2.
- Now, teacher will ask students to solve 'Check Point 2'.

- Teacher will define molecule, molecule of an element, molecule of a compound, i.e., monoatomic, diatomic, triatomic, tetratomic and polyatomic and formula of a compound, showing the number of elements in some compounds.
- Teacher should ask the students to solve 'Check Point 3'.
- At last, teacher will sum up the lesson by going through the points given under the head 'Wrapping It Up'.
- Teacher will finally help students to answer the questions given under the head 'Test Yourself'.

Boost Up

- Teacher can make students revise new terms given under the head 'Know These Terms'.
- Teacher can encourage students to learn the facts given under the head 'Something More'.
- Teacher can show animations related to the topics taught, if possible.
- Teacher should call each student one-by-one to tell the definition of matter and properties of a pure substance.
- Students should be asked to write few elements and also two metals and their properties.
- Students should be asked to remember the name and symbol of some elements given in Table 2.3 and Table 2.4.
- Students should be asked to tell one metal and one nonmetal behaving as liquid at room temperature.
- Students should be asked one-by-one to write one metalloid on the blackboard, and to tell the charge present in proton, neutron and electron.
- The teacher should ask the students to write the symbol of few elements on the blackboard. He/She should also ask the students to answer the questions related to constituents of compounds, their properties; molecule and molecule of an element, molecule of a compound, formula of a compound; elements present in the compound.

Expected Learning Outcomes

The students understand and know:

- definition of matter.
- pure and impure substances.
- elements, their types, i.e., metals, nonmetals and their properties, metalloids and noble gases.
- atoms and symbols of elements.
- compounds and their properties.
- molecule and molecule of an element.
- formula of a compound.

Evaluative Questions

The teacher should ask the following questions to evaluate the students.

- **1.** What is a pure substance? Write its different types.
- **2.** Is mixture an impure substance?
- **3.** Name one liquid nonmetal.
- 4. Define metalloid.
- 5. Write the names of two noble gases.
- 6. Mention the charges present in all three subparticles.
- 7. Name the elements present in sodium chloride.
- 8. Name the molecule made up of many atoms.